

1) Pd₂(dba)₃, MTBE, **A**
RT (87%)

C

1) name of the reaction?

2) Pd(Cl)₂, Cu(OAc)₂, H₂O, DMA, O₂ (63%);

3) KOH, xylenes, 110 °C (92%)

D

2) name of the reaction

4) TMS-acetylene, UV-B lamps, acetone

5) BF₃·OEt₂, DCM
6) TBAF, THF (68%, o3s)

E

Without step 5 yield is only 45%. How you could explain this?

HINT:
in step 4) also some rearrangement takes place also name this!

7) Pd(P(*t*Bu)₃)₂, **B**, NaO*t*Bu, THF,

MW, 120 °C (67%)

F

step 7) explain stereochemistry

8) Br₂, CHCl₃, (65%);
9) MW, *o*-dichlorobenzene, 250 °C, 3h (68%)



K

15) Pd/C, H₂ (97%)

16) NOBF₄, MeCN (70%)

step 16) exact mechanism?
HINT: N₂O is formed

J

14) LDA (3 equiv.)

THF, -20 °C (83%)

step 14) what intermediate is formed????

I

Compound I.
Try to explain observed stereochemistry

12) LDA, THF; MeI (68%);

13) DIBAL, PhMe (91%)

H

step 10) What is the name of catalyst?

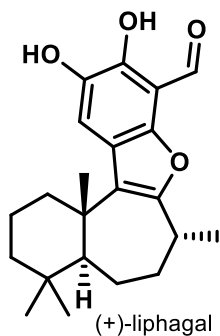
10) PtO₂, H₂ (69%);

11) NaOMe, MeOH 65 °C, 4 days (78%, 3 cycles)

G

step 9) what is the type of reaction?

17) *n*BuLi, TMEDA; DMF (70%)
18) BI₃, DCM (45%)



You already guessed. Propose mechanism to compound **M**.

