

ADVANCED FUNCTIONAL MATERIALS

Supporting Information

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Ligand-Programmed Consecutive Symmetry Break(s) in
Nanoparticle Based Materials Showing Emergent Phenomena:
Transitioning from Sixfold to Threefold Symmetry in
Anisotropic ZnO Colloids

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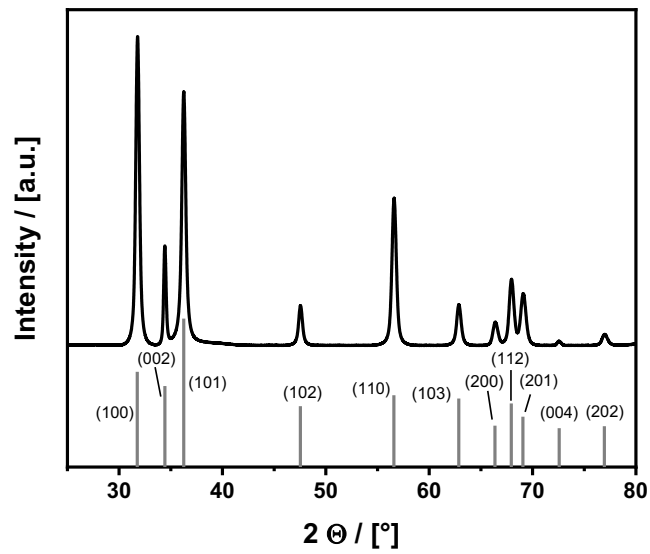
(d) Dr. U. Krings

Leibniz-University of Hannover, Institute of Food Chemistry, Callinstrasse 3-9, 30167 Hannover, Germany

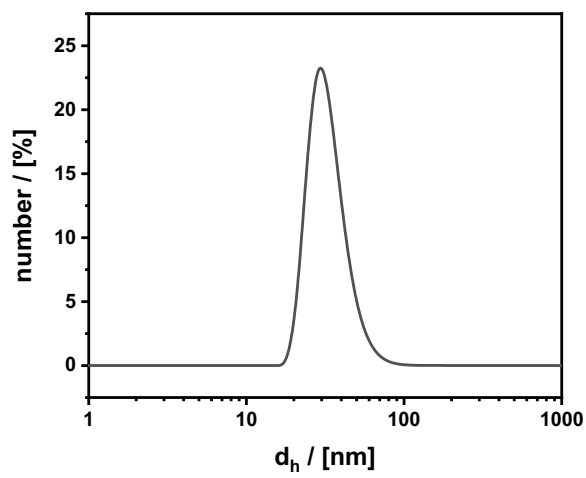
ELECTRONIC SUPPLEMENTARY INFORMATION

Fig. S1:

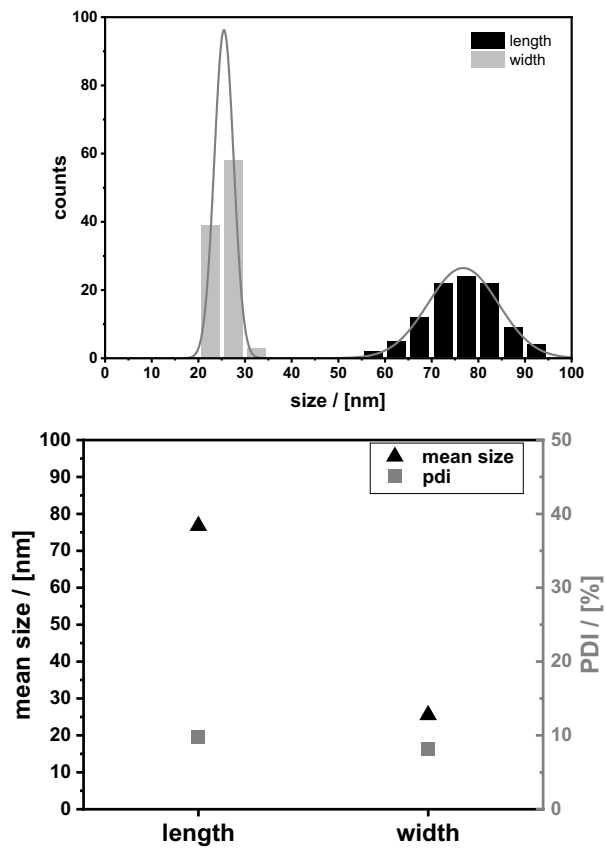
a) PXRD measurement of the trigonal ZnO nanoparticles



b) DLS Measurement of a dispersion of the trigonal ZnO nanorods in cyclohexane



c) Evaluation of polydispersity and aspect ratio of trigonal ZnO nanorods



d) UV-Vis spectra of triangular ZnO nanorods

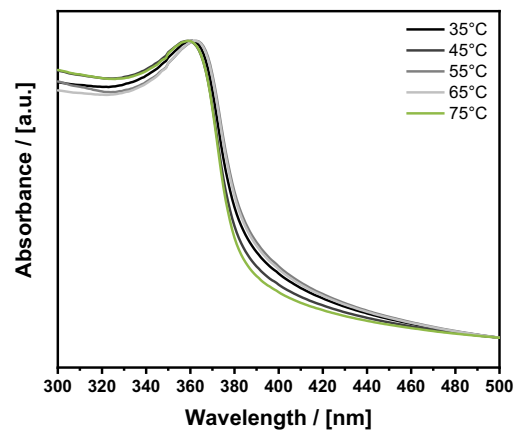


Fig- S2:

Lewis-Structures of Polyglyceryl-3-polyricinoleate (P3P) as well as Glycerol-monooleate (Mono-OI), 1,3-Glycerol-dioleate (Di-OI) and Glycerol-trioleate (Tri-OI):

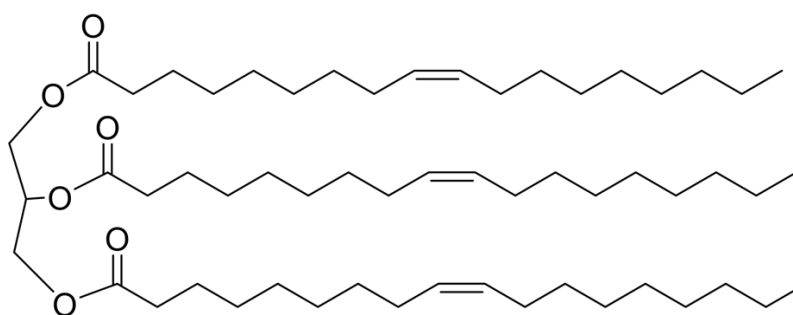
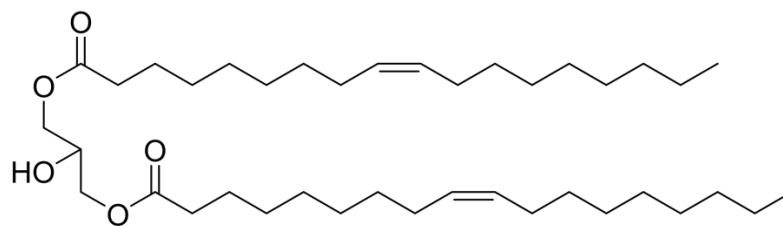
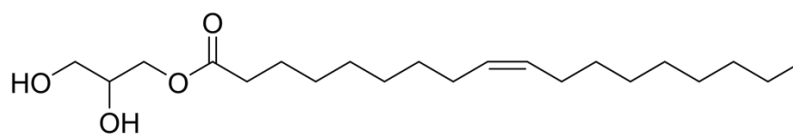
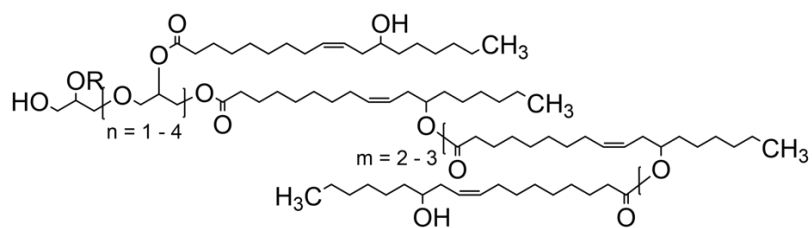
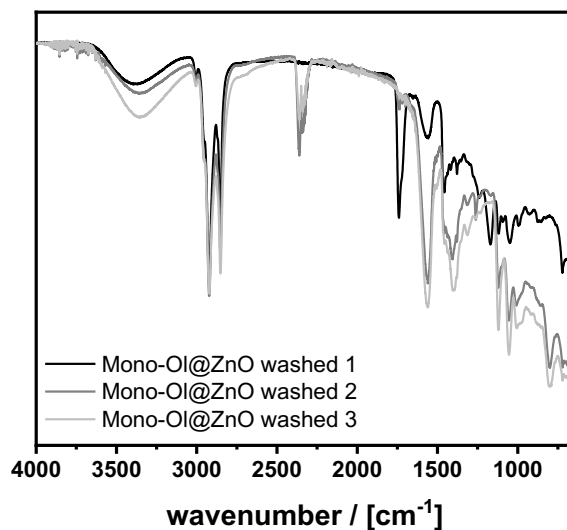


Fig. S3:

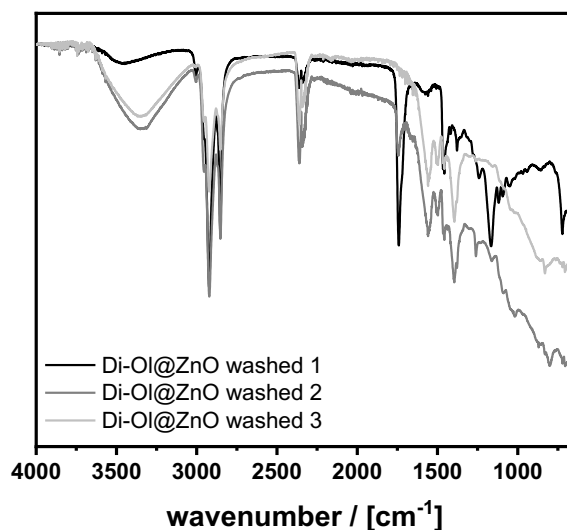
Normalized IR spectra of ZnO nanoparticles synthesized with pure Mono-OI a), pure Di-OI b), and pure Tri-OI c), each for three identical washing cycles:

a)



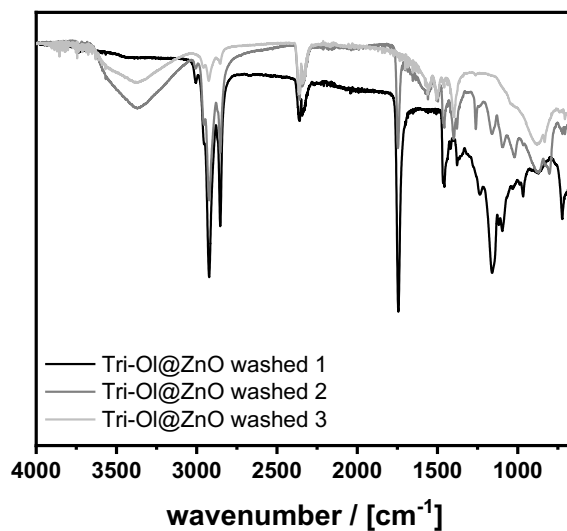
~**3500-3200 cm⁻¹**: O-H stretching bond of residual water; **3000-2800 cm⁻¹**: aliphatic C-H stretching bond; ~**2300 cm⁻¹**: C=C stretching bond; **1739 cm⁻¹**: non-coordinated carbonyl stretching bond, C=O; **1560 cm⁻¹**: shifted, coordinated carbonyl stretching bond; **1400 cm⁻¹**: asymmetric C-H bending; **1300-1000 cm⁻¹**: aliphatic C-O ester stretching bond.

b)



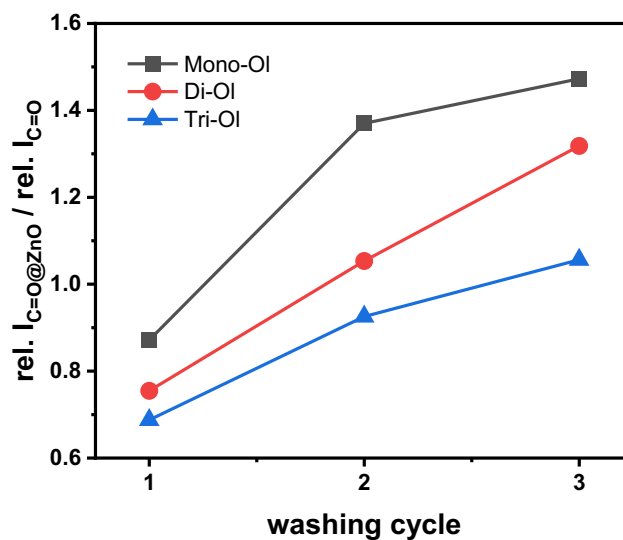
~**3500-3200 cm⁻¹**: O-H stretching bond of residual water; **3000-2800 cm⁻¹**: aliphatic C-H stretching bond; ~**2300 cm⁻¹**: C=C stretching bond; **1740 cm⁻¹**: non-coordinated carbonyl stretching bond, C=O; **1557 cm⁻¹**: shifted, coordinated carbonyl stretching bond; **1394 cm⁻¹**: asymmetric C-H bending; **1300-1000 cm⁻¹**: aliphatic C-O ester stretching bond.

c)



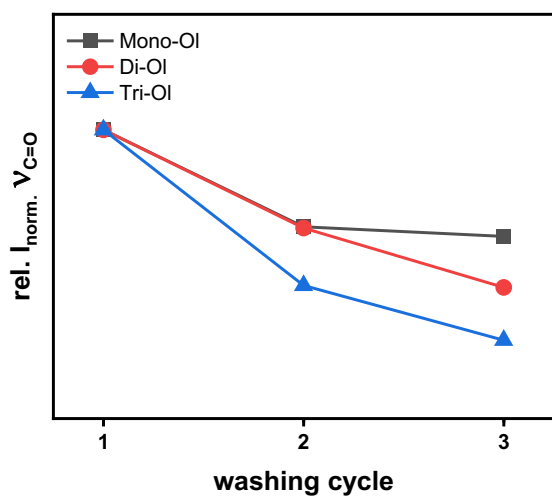
~3500-3200 cm⁻¹: O-H stretching bond of residual water; 3000-2800 cm⁻¹: aliphatic C-H stretching bond; ~2360 cm⁻¹: C=C stretching bond; 1743 cm⁻¹: non-coordinated carbonyl stretching bond, C=O; 1560 cm⁻¹: shifted, coordinated carbonyl stretching bond; 1394 cm⁻¹: asymmetric C-H bending; 1300-1000 cm⁻¹: aliphatic C-O ester stretching bond.

d)

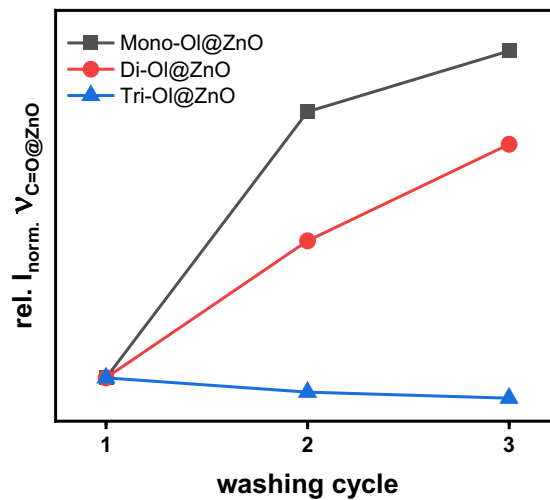


Ratios of the normalized relative intensities of coordinated and non-coordinated C=O stretching bonds of all three ligands (Mono-, Di-, and Tri-OI), each for three identical washing cycles:

e)



f)

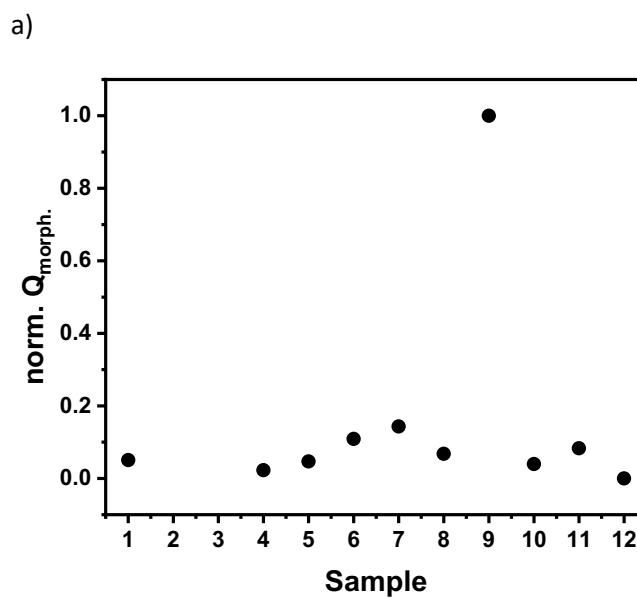


Normalized relative intensities of the free and coordinated C=O-stretching bonds of the three ligands Mono-OI, Di-OI and Tri-OI, each for three identical washing cycles:

Table S1: Molar ratios of the different glycerol ligands and corresponding values of Q_{morph} .

Sample	$X_{\text{Mono-OI}}$ [mol%]	$X_{\text{Di-OI}}$ [mol%]	$X_{\text{Tri-OI}}$ [mol%]	Q_{morph} .
1	100	0	0	0.05037
2	0	100	0	--
3	0	0	100	--
4	33.3	33.3	33.3	0.02255
5	50	50	0	0.04685
6	80	20	0	0.10877
7	70	30	0	0.14336
8	65	27.85	7.15	0.06765
9	42	58	0	1
10	40	17.15	42.85	0.0397
11	40	45	15	0.08307
12	30	70	0	0

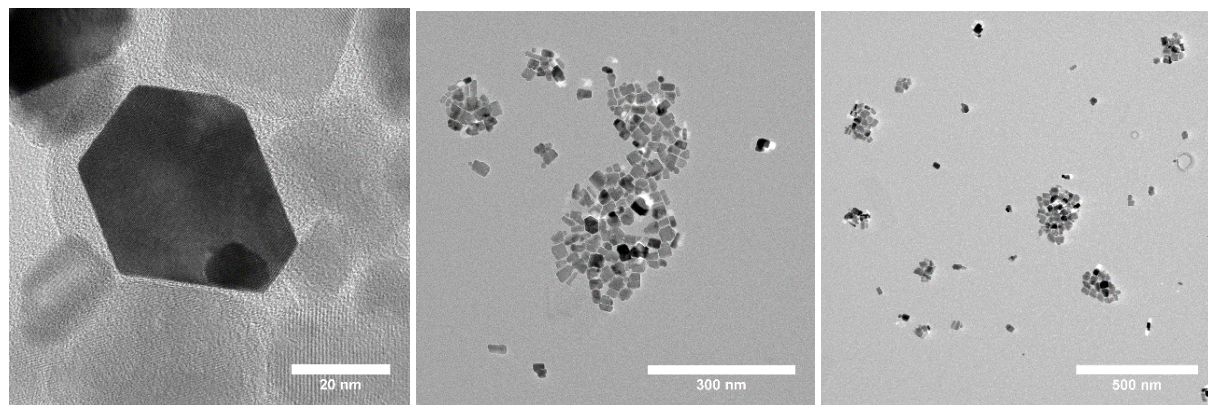
Fig S4:



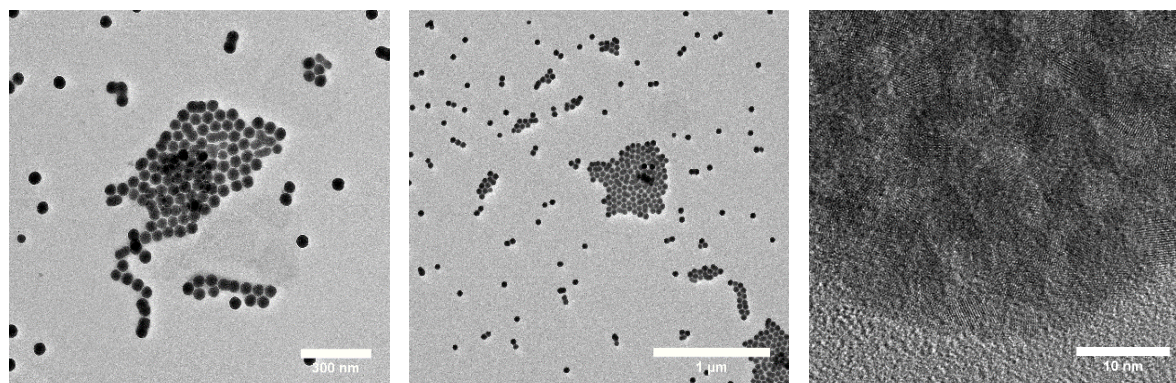
Normalized plot of $Q_{\text{morph.}}$ of the different molar compositions of the glycerol ligands presented in Table S1.

b) TEM images of ZnO samples, synthesized with different compositions of glycerol ligands as presented in Table S1.

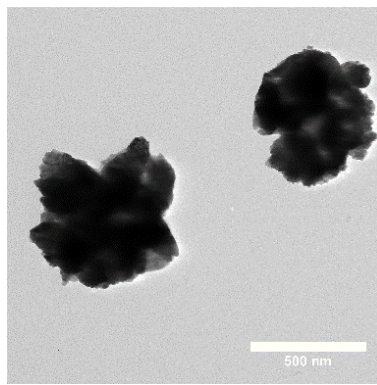
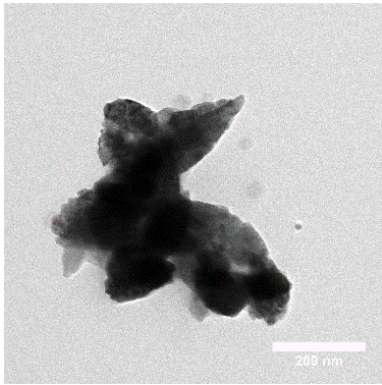
Sample 1:



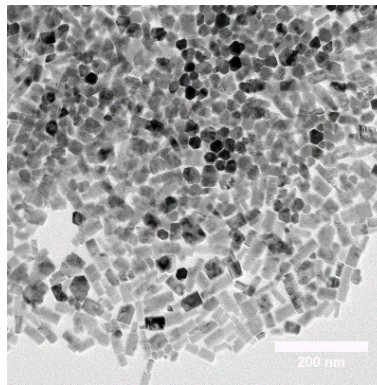
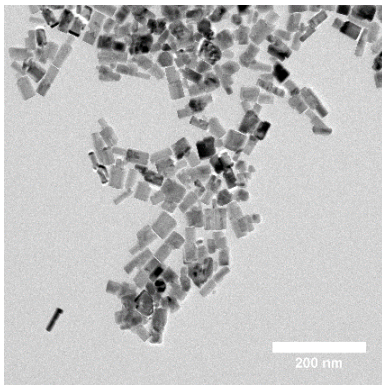
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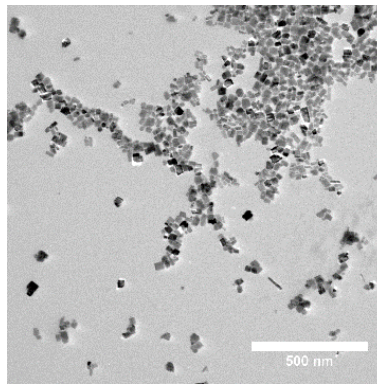
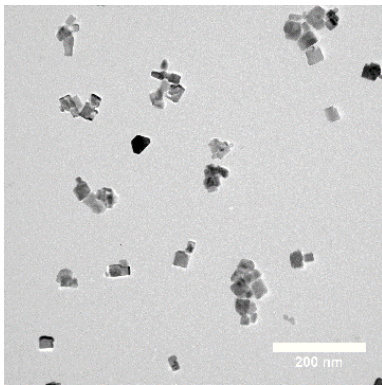
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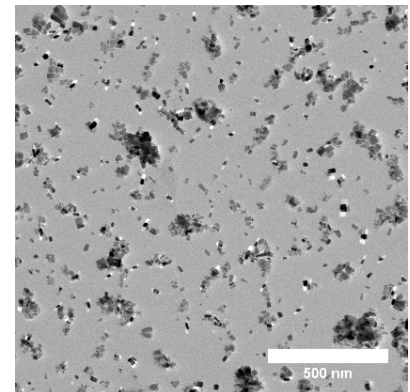
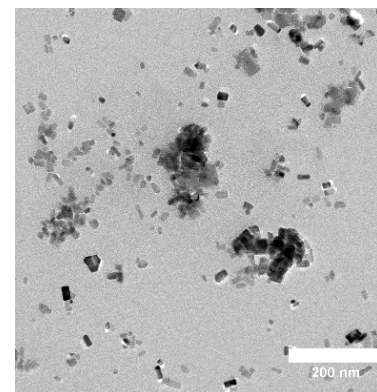
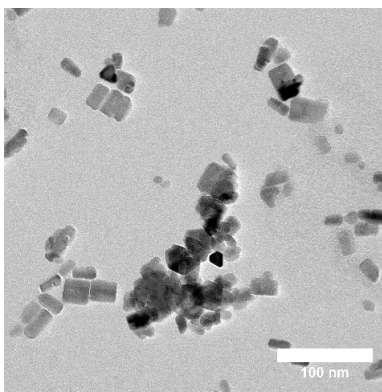
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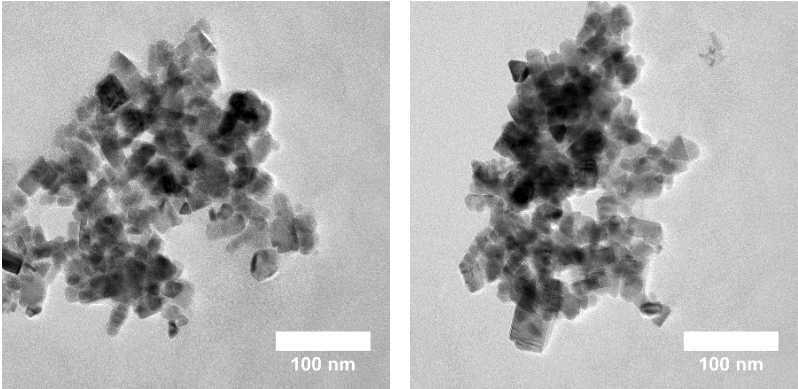
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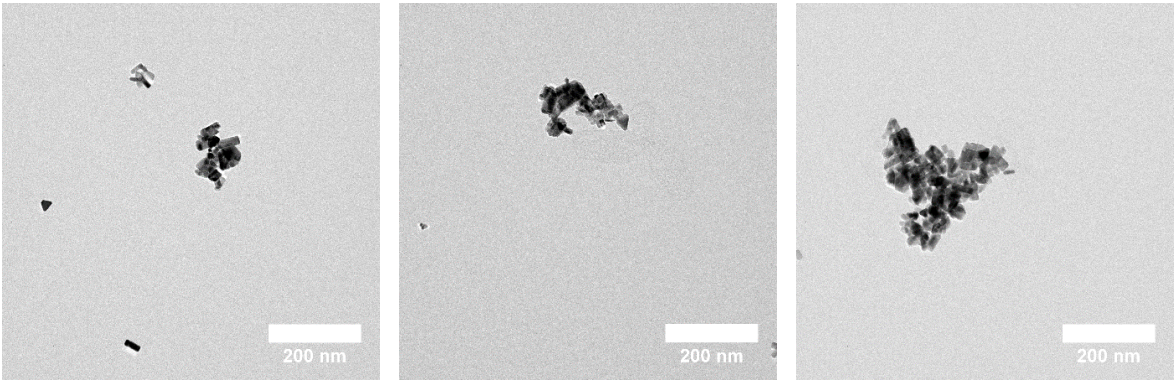
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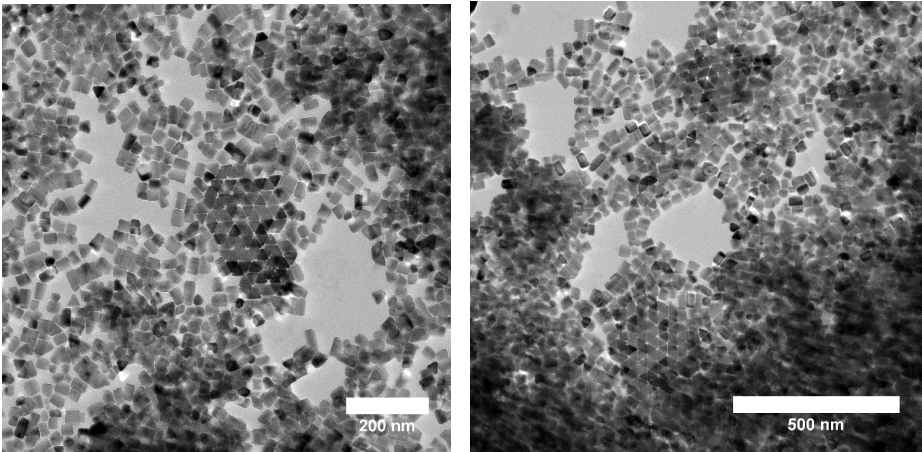
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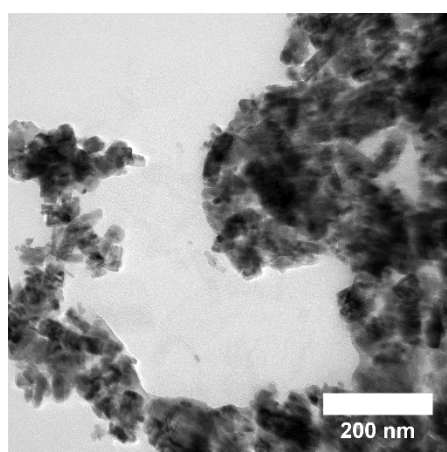
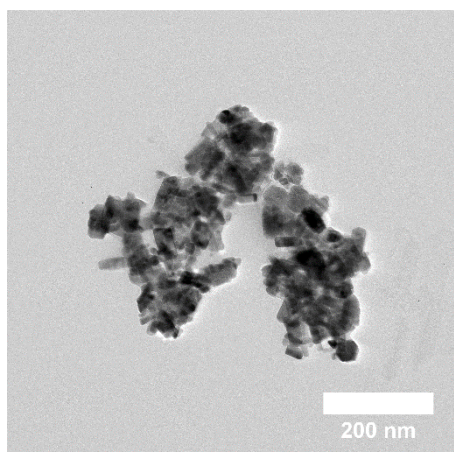
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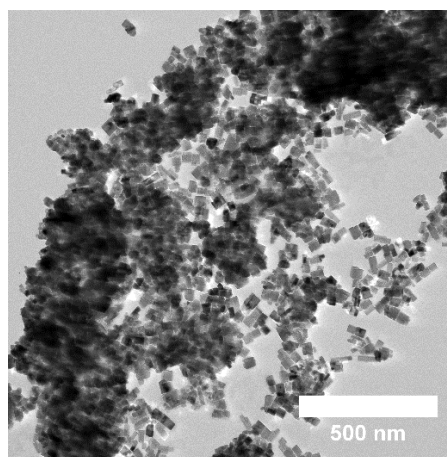
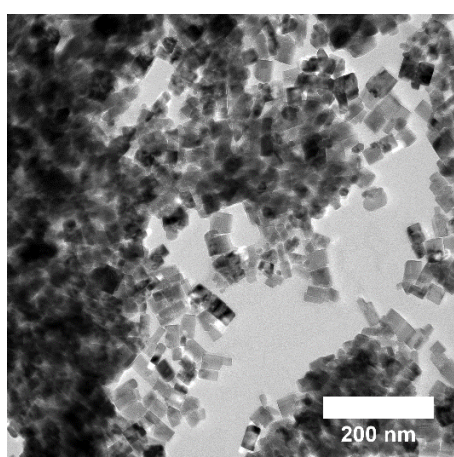
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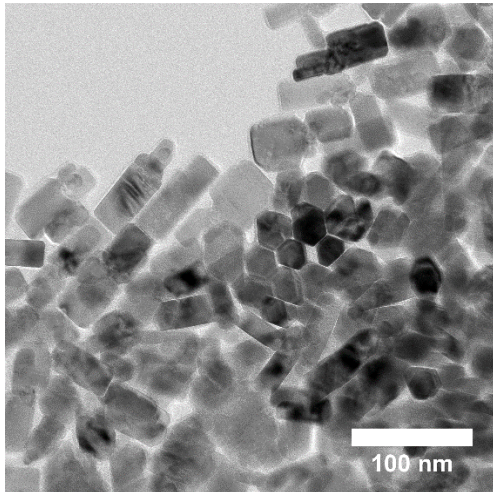
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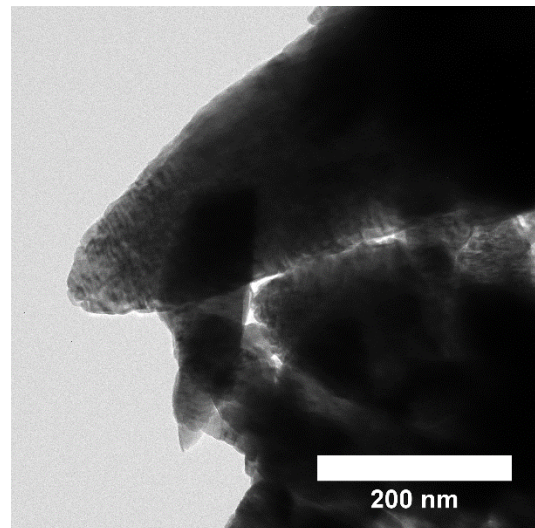
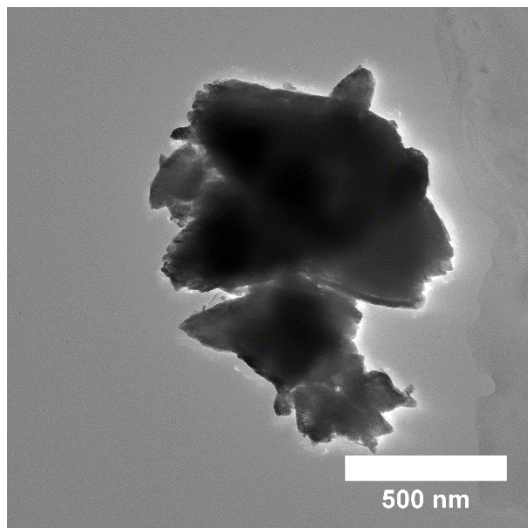
Sample 11:



Sample 12:



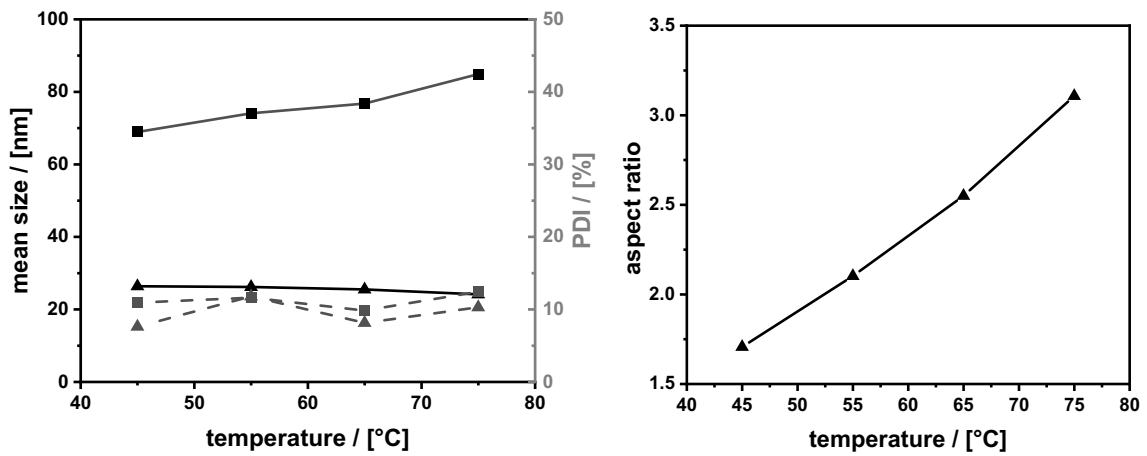
c)



TEM images of ZnO particles, synthesized without any of the three ligands.

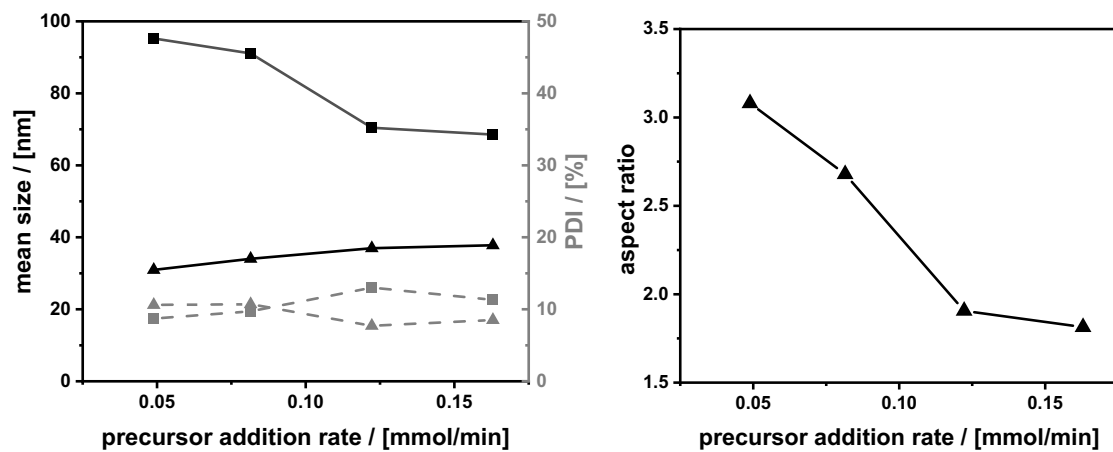
Fig S5:

a)



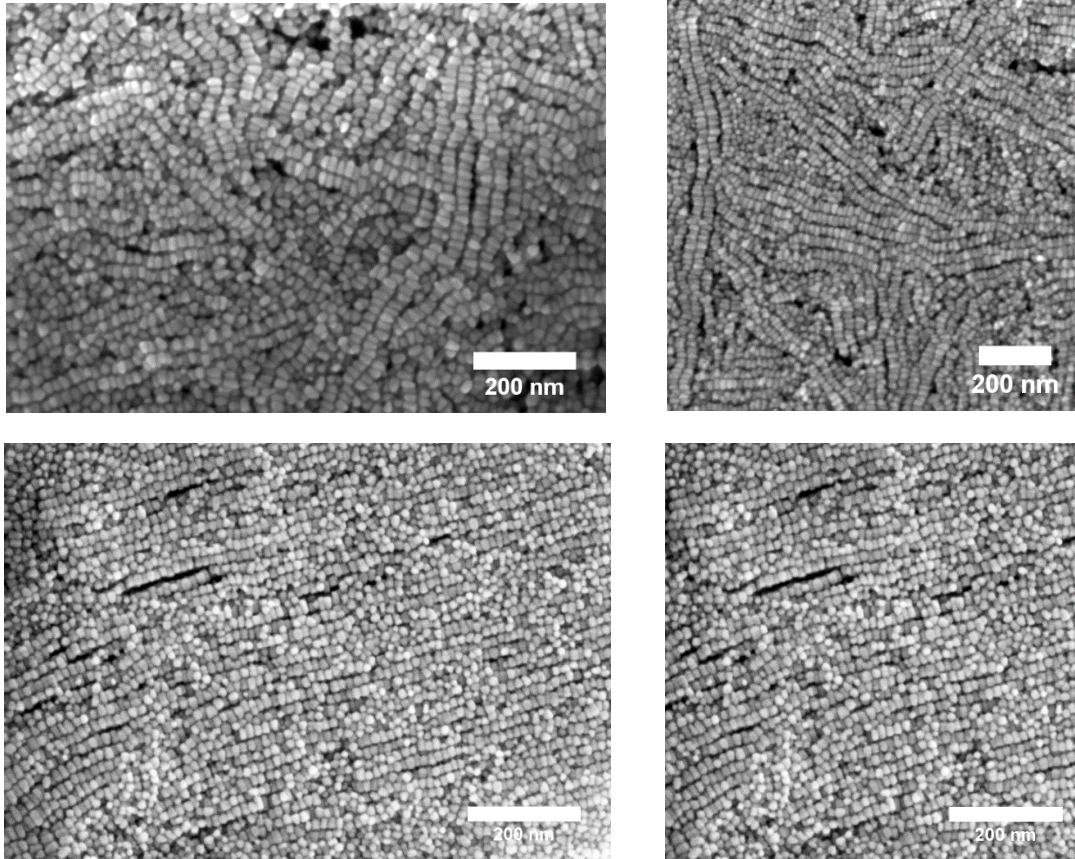
Influence of the temperature on the dimensions and the polydispersity of the nanoparticles during the synthesis; squares implicate the length (c), triangles implicate the width (a, b) of the nanoparticles.

b)



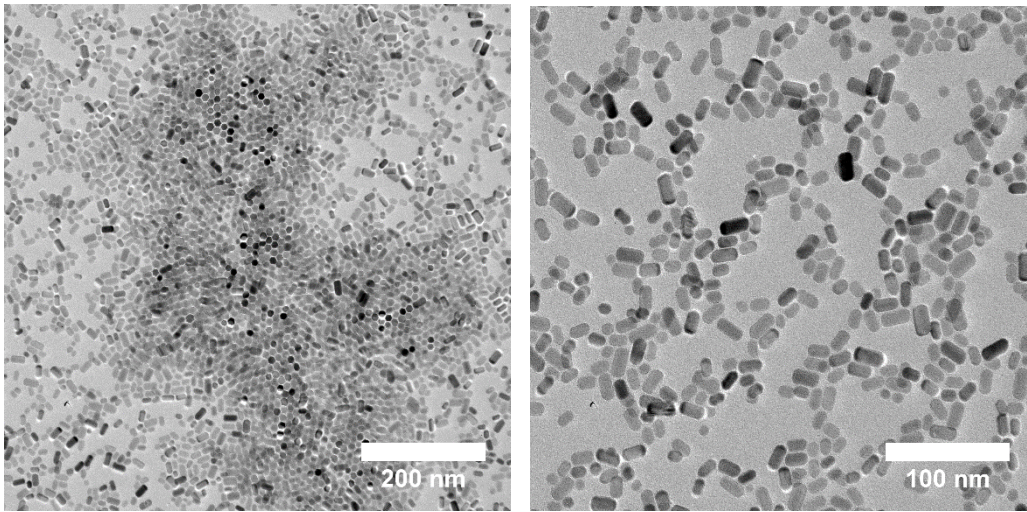
Influence of the precursor addition rate on the dimensions and the polydispersity of the nanoparticles during the synthesis; squares implicate the length (c), triangles implicate the width (a, b) of the nanoparticles.

Fig. S6:



Side-view SEM images of superstructures of ZnO nanoparticles. The particles form ladder-like arrays spanning over several hundred nanometres.

Fig. S7:



TEM images of ZnO nanorods (hexagonal).